

Crystal Chemistry of α -AlB₁₂ and γ -AlB₁₂

Iwami Higashi

Department of Chemistry, Chiba Institute of Technology, Shibazono, Narashino, Chiba 275-0023, Japan

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The structures of α -AlB₁₂ and γ -AlB₁₂ are both characterized by a rigid boron framework derived from a combination of three-dimensional networks of icosahedral B₁₂ and B₂₀ units. Crystals of γ -AlB₁₂ (*P*2₁2₁2; *a* = 16.57, *b* = 17.51, *c* = 10.14 Å) are almost always obtained as syntactically intergrown crystals with α -AlB₁₂ (*P*4₃2₁2 or *P*4₃2₁2; *a* = 10.16, *c* = 14.27 Å) from high-temperature Al–B solutions. Examination of structural details of α -AlB₁₂ and γ -AlB₁₂ demonstrated that there are close similarities between them, notwithstanding marked differences in the crystal data. The syntactic intergrowth between α -AlB₁₂ and γ -AlB₁₂ is always effected so as to have the (101)_{*x*} and the (100)_{*y*} planes parallel to each other and the *a_x* or *b_x* axis and the *c_y* axis in the same direction; these planes are identical in unit area and in the orientation inside the intergrown crystals. This fact, as well as the fact that the icosahedral B₁₂ networks of both crystals which are parallel to the syntactic intergrowth plane are identical, may facilitate the formation of the intergrown crystal. Structural details of one crystal are compared to the corresponding ones of the other. © 2000 Academic Press

1. INTRODUCTION

At present in the binary Al–B system only three phases, AlB₂ (1), α -AlB₁₂ (2, 3), and (with less confidence) γ -AlB₁₂ (3), are considered to be pure binary aluminum borides. The phases AlB₁₀ (4) and β -AlB₁₂ (2) were shown to be ternary compounds containing carbon. It was reported by Matkovich *et al.* (5) that γ -AlB₁₂ could not be prepared as single crystals but as intergrown crystals with α -AlB₁₂.

By use of high-temperature Al–B solution growth methods or by sintering powdery Al–B compacts (6), we have also collaborated that as for boron-rich compounds only α -AlB₁₂ and γ -AlB₁₂ could be prepared, and these two boron-rich phases are always obtained as syntactically intergrown crystals as had been reported by Kohn and Eckart (3). Furthermore, our extensive efforts (6) to find a reproducible method to prepare the single phase or single crystals of γ -AlB₁₂, which eventually proved to be in vain, showed the somewhat unstable nature of this phase; when the intergrown crystals are pulverized to prepare a sample for powder X-ray diffraction, the γ -AlB₁₂ phase is entirely

transformed into the α -AlB₁₂ phase (6). Single-phase crystals of the γ -AlB₁₂ structure type doped with Mg (Al_{1.4}Mg_{0.5}B₂₂), which are obtainable by cooling a high-temperature Al–B solution containing a small quantity of Mg, however, cannot be transformed into the α -AlB₁₂ structure type by pulverization (6). Crystal data for α -AlB₁₂, and γ -AlB₁₂ phases are presented in Table 1.

The α -AlB₁₂ phase was prepared as pure binary crystals by cooling Al–B melts from temperatures of about 1500°C, and its structure determined by the single-crystal X-ray diffractometry (8, 9). Since the structure analysis of γ -AlB₁₂ was done by using the γ -AlB₁₂ crystals which had been prepared by the same method as for the α -AlB₁₂ phase (possibly at slightly lower temperatures) (10, 11), it would be highly probable that the crystals used for the X-ray diffraction work had contained a small quantity of Mg or certain other impurity atoms. From this context, it is of interest to point out that one set of the published data (*a* = 16.623(5), *b* = 17.540(5), *c* = 10.180(5) Å) of the γ -AlB₁₂ single crystal utilized for the structure determination (10) is almost the same as those of the Al_{1.4}Mg_{0.5}B₂₂ crystal of the γ -AlB₁₂ structure type (7), although they are significantly different from those for another γ -AlB₁₂ crystal (11) (see Table 1); the latter had been found by chance among a large amount of the intergrown crystals of α -AlB₁₂ and γ -AlB₁₂ (12). Although there are significant differences in the metal distribution, the results from three publications (7, 10, 11) from which the crystal data are cited in the Table 1 were all the same with regard to the structure of boron framework.

In the present paper structural details of boron frameworks in the α -AlB₁₂ and the γ -AlB₁₂ structures are described, demonstrating that there are close similarities between the two structures, notwithstanding a great difference in the crystal data (Table 1).

2. DESCRIPTIONS OF STRUCTURES

Structural data are available from Refs. (8 and 9) for α -AlB₁₂ and from Refs. (10 and 11) for γ -AlB₁₂. Although the structures of α -AlB₁₂ (8) and γ -AlB₁₂ (11) have been described in detail, some of their general features will be given

TABLE 1
Crystal Data for the Crystals of α -AlB₁₂ and γ -AlB₁₂
Structure Type

| Crystal | α -AlB ₁₂ | γ -AlB ₁₂ | γ -AlB ₁₂ | Al _{1.4} Mg _{0.5} B ₂₂ ^a |
|-------------------------------|-----------------------------------|-----------------------------------|-----------------------------------|--|
| Crystal system | Tetragonal | | Orthorhombic | |
| <i>a</i> (Å) | 10.158(2) | 16.573(4) | 16.623(5) | 16.633(3) |
| <i>b</i> (Å) | (= <i>a</i>) | 17.510(3) | 17.540(5) | 17.547(4) |
| <i>c</i> (Å) | 14.270(5) | 10.144(1) | 10.180(5) | 10.187(2) |
| Chemical formula ^b | B ₂₂ Al _{1.6} | B ₈₈ Al _{6.3} | B ₈₈ Al _{6.5} | B ₈₈ Al _{5.6} Mg ₂ |
| Reference | 8 | 11 | 10 | 7 |

^a γ -AlB₁₂ structure type.

^b Asymmetric units determined by x-ray structure analysis; expressed in terms of boron structural subunits, they are 1/2 (2B₁₂·B₂₀(I)) and B₁₂(I)·B₁₂(II)·B₁₂(III)·B₁₂(IV)·B₂₀(I)·B₂₀(II) for α -AlB₁₂ and γ -AlB₁₂ structure type, respectively.

here to facilitate discussions. There are three main components of the structures of α -AlB₁₂ and γ -AlB₁₂: (i) arrangements of icosahedral B₁₂ units; (ii) arrangement of B₂₀ units; and (iii) statistical distribution of metal atoms in the boron framework. The B₂₀ unit is a twinned icosahedron with a B–B–B triangular composition plane. It has two vacant apical sites and an isolated B atom bridging both sides of the unit. There are two types of B₂₀ unit; stereoscopic views of them are presented in Figs. 1 and 2. One of the B₂₀ units, B₂₀-(C₂), having one vacant site at each side of the unit (Fig. 1), is that which is present in the α -AlB₁₂ structure, and the other, B₂₀-(C_s), having two vacant sites at one side of the unit (Fig. 2), is the one that has been found only in γ -AlB₁₂; in the γ -AlB₁₂ structure both types of B₂₀ units are present (10). Each of the symbols characterizes the symmetry of the unit in the standard Schoenflies notation (10). In earlier papers, the B₂₀ unit in α -AlB₁₂ (8, 9) and the B₂₂ unit in β -tetragonal boron (13) were referred to as the B₁₉ unit and the B₂₁ unit, respectively, excluding the isolated B atom bridging both sides of the twinned icosahedron; the B₂₂ unit is identical with the B₂₀ unit except that it has no vacant apical site. From an electronic point of view (11, 14, 15), however, the superficially isolated

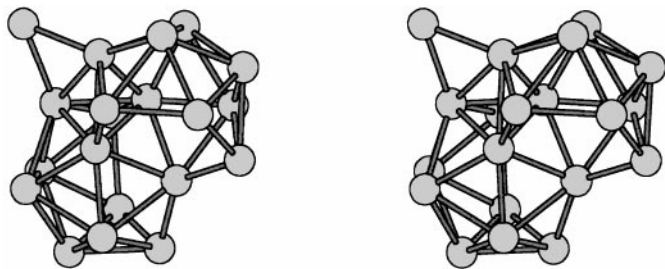


FIG. 1. Stereoscopic view of B₂₀-(C₂) unit.

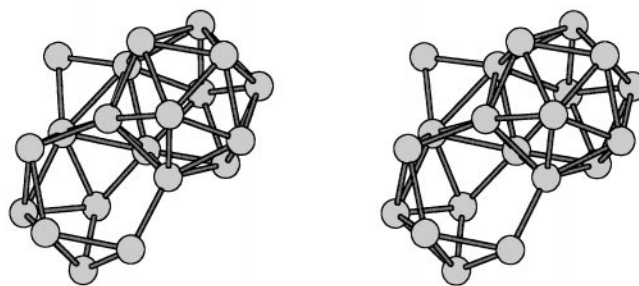


FIG. 2. Stereoscopic view of B₂₀-(C_s) unit.

single B atom bridging two sides of each of the units is regarded as an additional member of the twinned icosahedron units (16).

In the structures of both α -AlB₁₂ and γ -AlB₁₂, the linkages of B₁₂–B₁₂, B₁₂–B₂₀, and B₂₀–B₂₀ are effected through quasi-fivefold axes of B₁₂ icosahedra and similar directions of the B₂₀ units.

2.1. α -AlB₁₂

The structure of α -AlB₁₂ (8, 9) is characterized by the presence of an icosahedral B₁₂ framework (Fig. 3) combined with that of B₂₀ units (Fig. 4). As mentioned above, the B₂₀ units in the α -AlB₁₂ structure are of the B₂₀-(C₂) type. The arrangement of B₁₂ icosahedra as presented in Fig. 3 is described as an alternate alignment of icosahedral B₁₂ chains running along the *a* and *b* axes; the alternate alignment is generated by the operation of the 4₁ or 4₃ screw axis (of the space group *P*4₁2₁2 or *P*4₃2₁2), which results in the construction of a rigid three-dimensional icosahedral B₁₂ framework having large openings capable of accommodating B₂₀ units. By rephrasing icosahedral B₁₂ chains with zigzag chains of B₂₀ units, a similar description can be made regarding the construction of a rigid framework of B₂₀ units (Fig. 4). The nature of the coordination of B₁₂ icosahedra about the B₂₀ unit is the same as that in the γ -AlB₁₂ structure, which will be described below. The α -AlB₁₂ phase decomposes by arc melting, expelling the Al metal to form a solid solution of the β -rhombohedral boron structure with a composition of AlB₃₁ (17); the decomposition temperature is around 2000°C (3, 18).

2.2. γ -AlB₁₂

As shown in Table 1, the asymmetric unit of boron in γ -AlB₁₂ consists of four crystallographically independent B₁₂ icosahedra, B₂₀-(C₂), and B₂₀-(C_s); on the other hand, that in α -AlB₁₂ is 1/2(2B₁₂·B₂₀). As in the case of the α -AlB₁₂ structure, the structure of γ -AlB₁₂ is also characterized by the presence of an icosahedral B₁₂ framework combined with that of B₂₀ units. An icosahedral B₁₂ network as

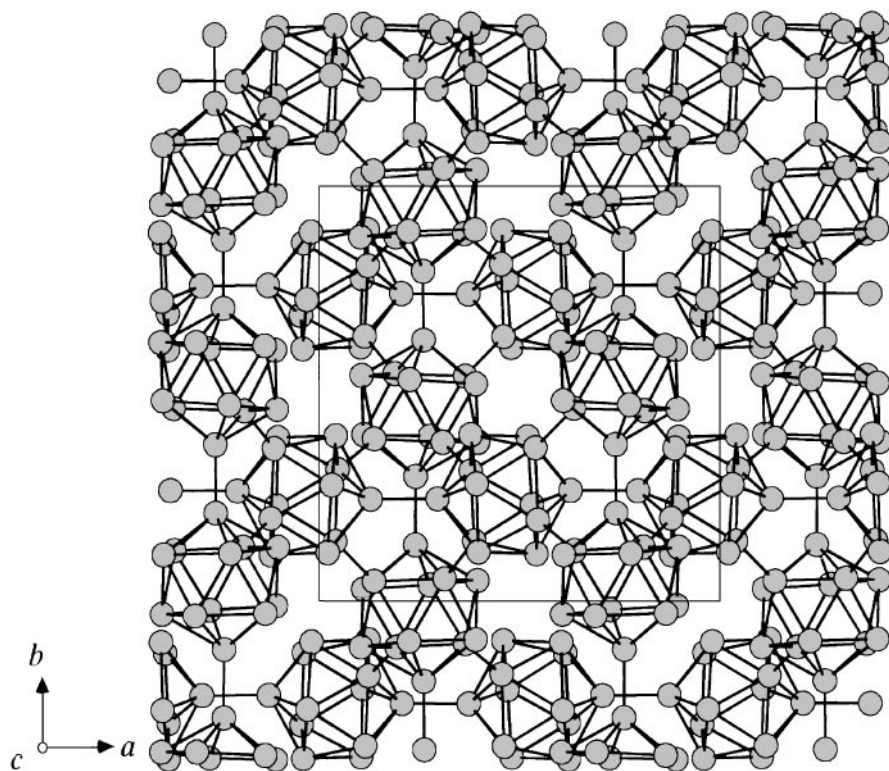


FIG. 3. Arrangement of B₁₂ icosahedra in the α -AlB₁₂ structure as seen along the *c* axis.

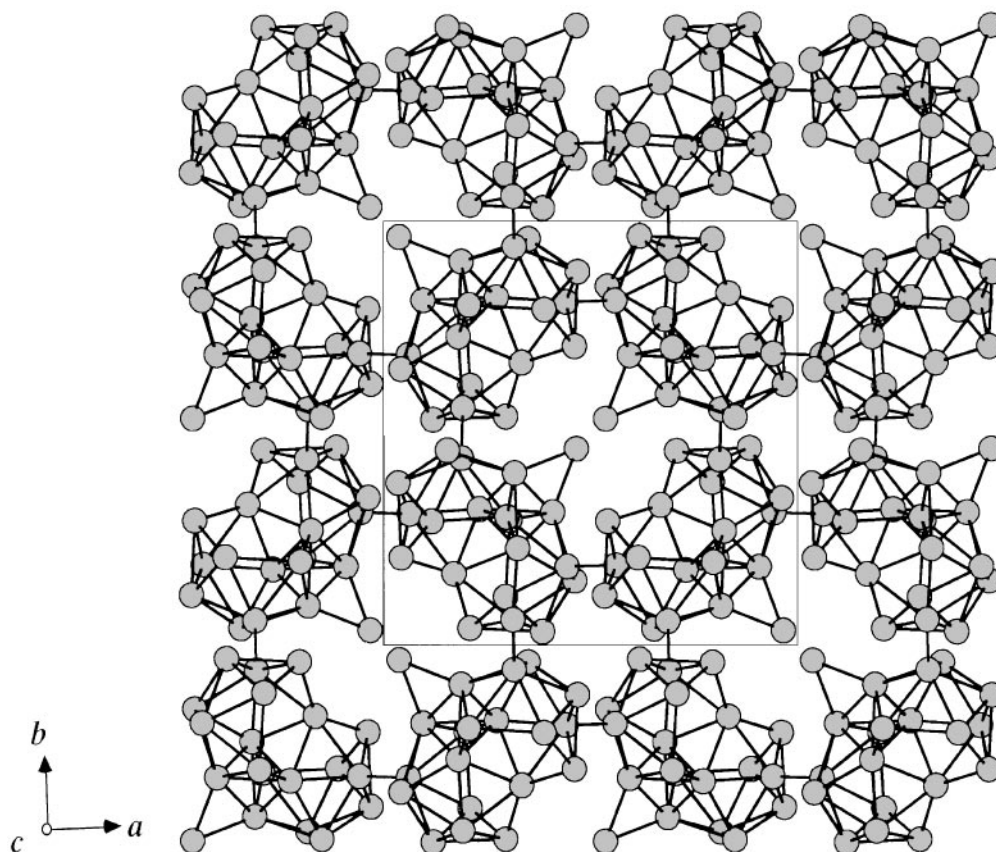


FIG. 4. Arrangement of B₂₀ units in the α -AlB₁₂ structure as seen along the *c* axis.

seen along the a axis is presented in Fig. 5, where four B₁₂ icosahedra are arranged in a tetrahedral form; the 4B₁₂ tetrahedron consists of the B₁₂ asymmetric unit in the γ -AlB₁₂ structure. By operating the twofold screw axis (of the space group $P2_12_12_1$), which lies at about the center of the 4B₁₂ tetrahedron in the direction of the a axis, the full structure of the icosahedral B₁₂ framework can be generated. As a result, two layers of the kagomé nets consisting of interconnected 4B₁₂ tetrahedra form the basic structure of the icosahedral B₁₂ framework. The large openings between two layers have the form of a truncated tetrahedron having a B₁₂ icosahedron at each of its 12 apical sites. The

B₂₀-(C₂) and B₂₀-(C_s) units fill the large openings, thus linking to 12 B₁₂ icosahedra arranged in the truncated tetrahedral form. Detailed descriptions of the coordinations of B₁₂ icosahedra surrounding each of the B₂₀ units are given in Ref. (11).

A boron network made up of the B₂₀-(C₂) and B₂₀-(C_s) units as seen along the a axis is presented in Fig. 6. It consists of B₂₀ zigzag chains running along the c axis direction. There are two kinds of zigzag chains; one is B₂₀-(C₂) and the other B₂₀-(C_s). Every B₂₀ unit within the zigzag chains is linked to a nearest B₂₀ unit of different zigzag chains, forming an approximately hexagonal

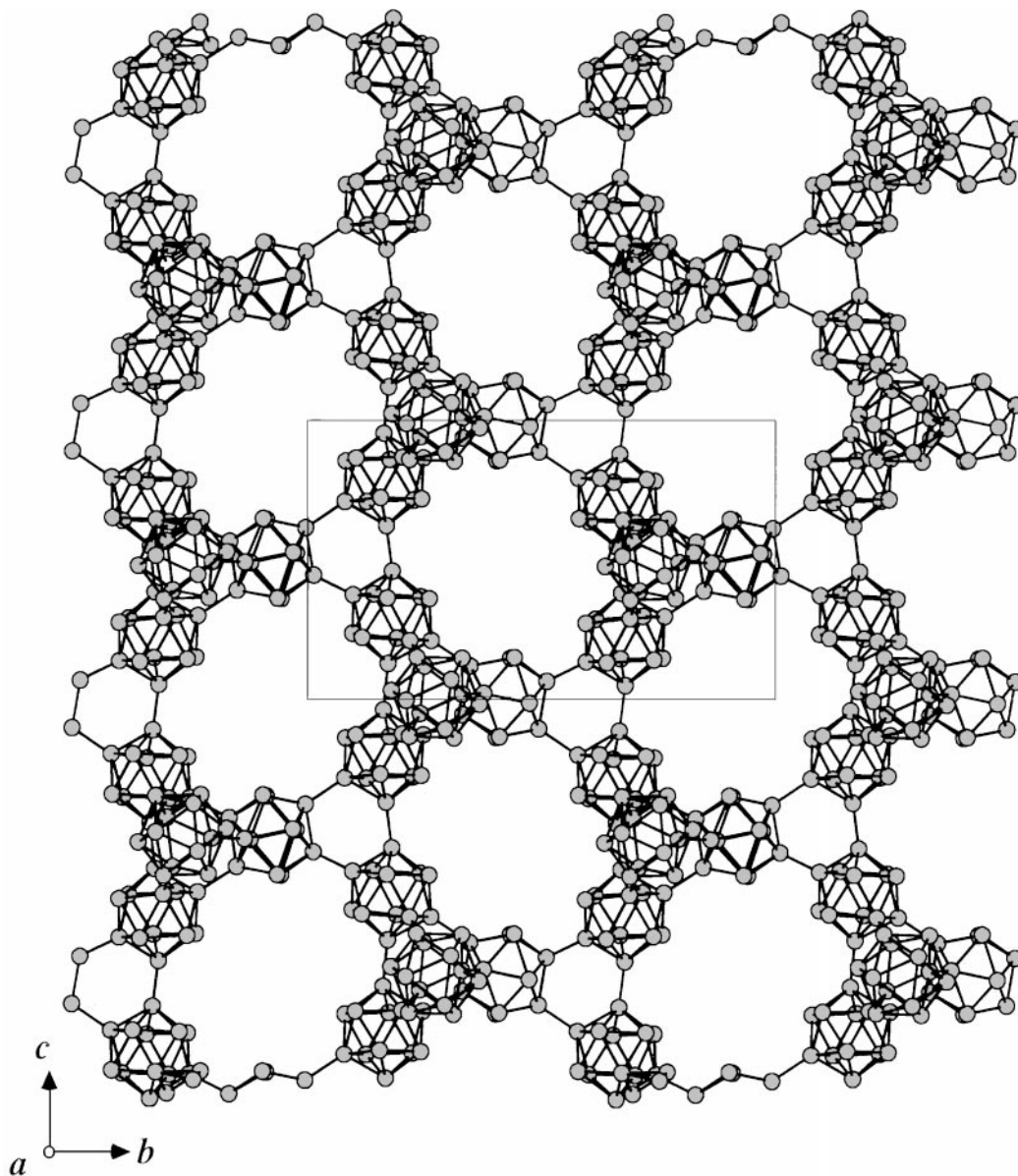


FIG. 5. Arrangement of B₁₂ icosahedra in the γ -AlB₁₂ structure as seen along the a axis.

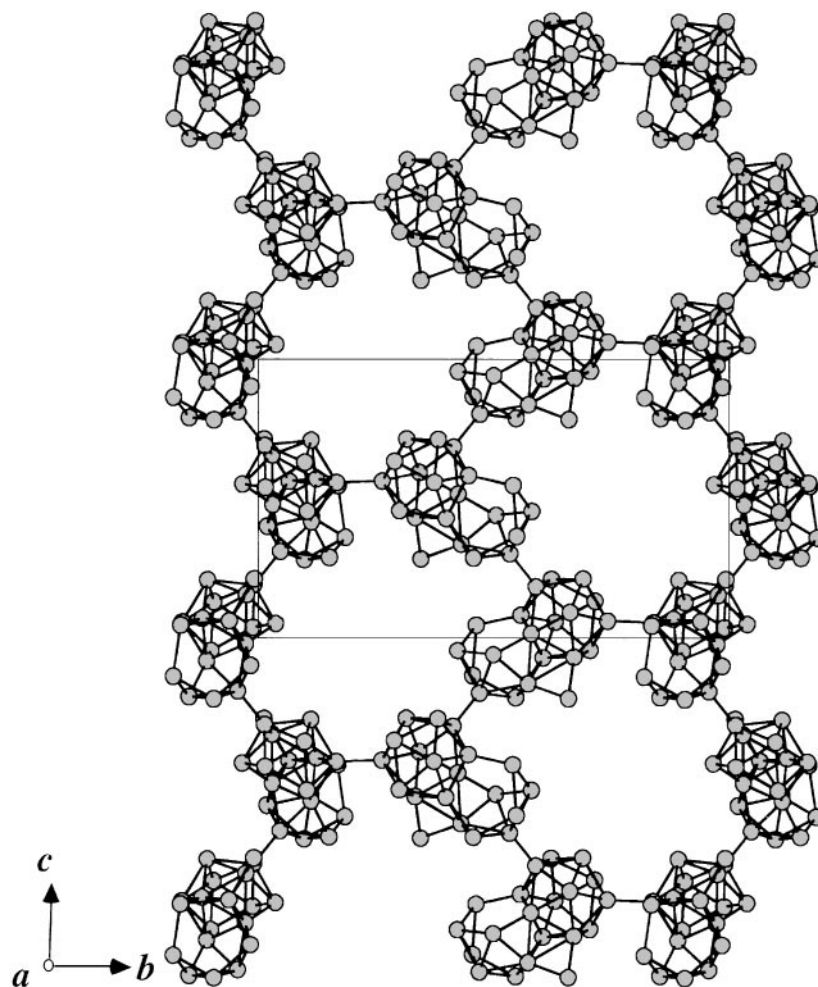


FIG. 6. Arrangement of B_{20} units in the $\gamma\text{-AlB}_{12}$ structure as seen along the a axis.

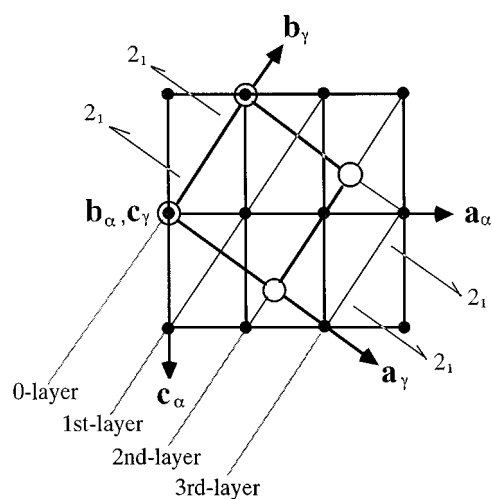


FIG. 7. Relationships of the unit cells or structures of $\alpha\text{-AlB}_{12}$ and $\gamma\text{-AlB}_{12}$ in the syntactically intergrown crystal of both phases. Solid black circles represent the lattice points of $\alpha\text{-AlB}_{12}$ and the larger white circles those of $\gamma\text{-AlB}_{12}$. The 2_1 symmetry element is related only to $\gamma\text{-AlB}_{12}$.

network of B_{20} units. As in the icosahedral B_{12} network, with the 2_1 symmetry axis laying at about the center of each hexagon of the network in the direction of the a axis, the full structure of the boron framework of B_{20} units is obtained. As can be expected from the figure, the linkages of the networks in the direction of the a axis occur between different types of B_{20} units just as in the b axis direction.

2.3. Aluminum Distributions in the Boron Frameworks of $\alpha\text{-AlB}_{12}$ and $\gamma\text{-AlB}_{12}$

The Al atoms are distributed over 5 sites ($\alpha\text{-AlB}_{12}$) or 11 sites ($\gamma\text{-AlB}_{12}$) which are located outside the B_{12} or B_{20} units. The occupancies of five sites in the $\alpha\text{-AlB}_{12}$ structure are respectively 0.72, 0.49, 0.24, 0.15, and 0.02 (8). In the $\gamma\text{-AlB}_{12}$ structure, the occupancies are 0.96, 0.95, 0.95, 0.76, 0.75, 0.52, 0.51, 0.44, 0.19, 0.19, and 0.07, respectively (11). From an electronic point of view, it would

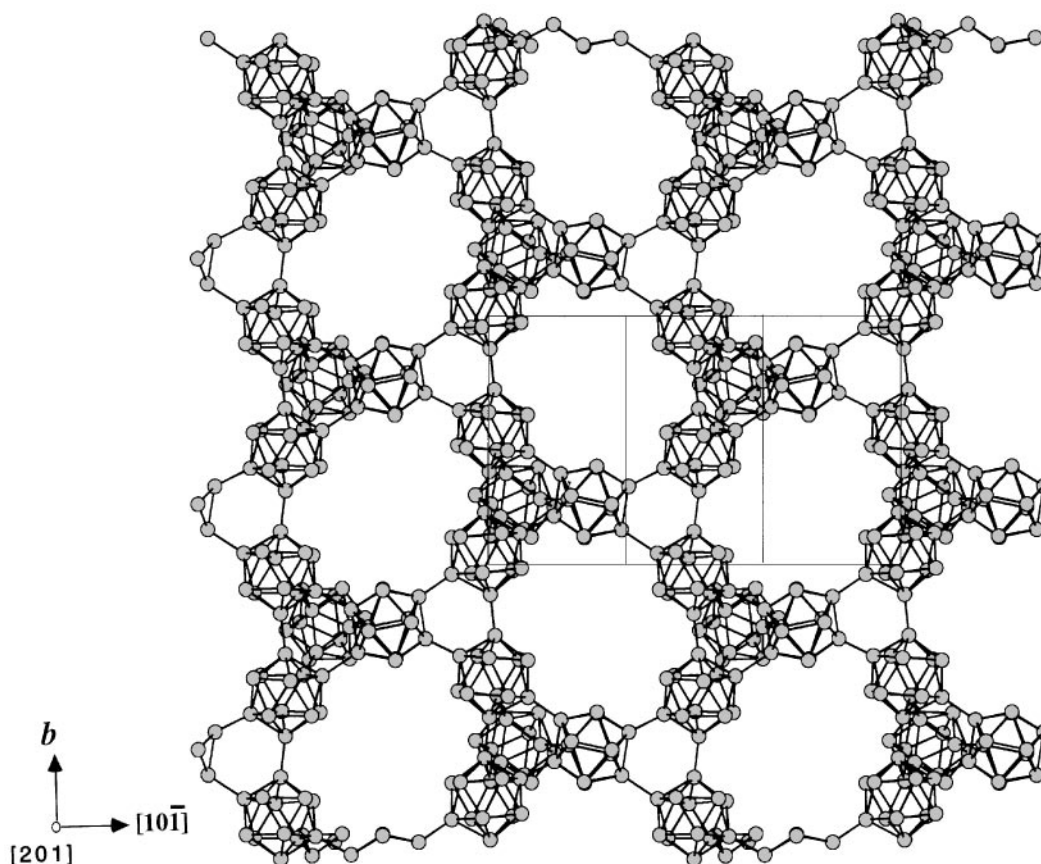


FIG. 8. Arrangement of B₁₂ icosahedra in the α -AlB₁₂ structure as projected along the (201) axis onto the (101) plane. The (101) unit plane indicated around the center of figure is divided into three equal parts, each of which corresponds to the d_{102} spacing.

be reasonable that the Al atoms are distributed so as to stabilize equally all the B structural subunits (B₁₂ and B₂₀), which are considered to be electron deficient. Therefore, charge balance among the boron structural subunits in α -AlB₁₂ and γ -AlB₁₂ was examined by counting the number of contacts between Al atoms and the boron units (11, 15). The result of the examination indicates that, although distributed in an apparently complicated manner, the Al atoms in the α -AlB₁₂ and γ -AlB₁₂ phases are distributed in such a way that they preserve a negative charge balance among B₁₂ and B₂₀ units in a ratio of about 1:3, which is probably essential to make up the stable boron frameworks. The results are compatible with the ionic formulas proposed from molecular orbital calculations (11, 14, 15): $(10/3) \text{Al}^{3+} \cdot 2\text{B}_{12}^{-2} \text{B}_{20}(\text{C}_2)^{6-}$ for α -AlB₁₂, and $(20/3)\text{Al}^{3+} \cdot 4\text{B}_{12}^{-2} \text{B}_{20}(\text{C}_2)^{6-} \text{B}_{20}(\text{C}_8)^{6-}$ for γ -AlB₁₂.

3. RELATIONSHIP OF THE STRUCTURES OF α -AlB₁₂ AND γ -AlB₁₂

As mentioned in the Introduction, the crystals of γ -AlB₁₂ are obtained as syntactically intergrown crystals with α -

AlB₁₂. The relationship of the unit cells of both phases in the syntactically intergrown crystals (4) is shown in the Fig. 7. The 2₁ symmetry element in the figure is applicable only to the γ -AlB₁₂ structure. In this figure, the unit cell of γ -AlB₁₂ is derived by the transformation of the α -AlB₁₂ unit cell as shown in the following:

$$\begin{pmatrix} a \\ b \\ c \end{pmatrix}_{\gamma} = \begin{pmatrix} 4/3 & 0 & 2/3 \\ 1 & 0 & -1 \\ 0 & 1 & 0 \end{pmatrix} \begin{pmatrix} a \\ b \\ c \end{pmatrix}_{\alpha}$$

The B₁₂ or B₂₀ frameworks of α -AlB₁₂ so far presented (Figs. 3 and 4) might seem quite different from those of γ -AlB₁₂ (Figs. 5 and 6). It is striking, however, that the B₁₂ or B₂₀ network of α -AlB₁₂ as projected along the (201) axis onto the (101) plane (Fig. 8 or 9) shows the same arrangement of the B₁₂ or B₂₀ units as that in the γ -AlB₁₂ structure (Figs. 5 and 6). The only difference between them is that in the α -AlB₁₂ structure the B₂₀ framework consists solely of B₂₀(C₂) units, while in the γ -AlB₁₂ structure the

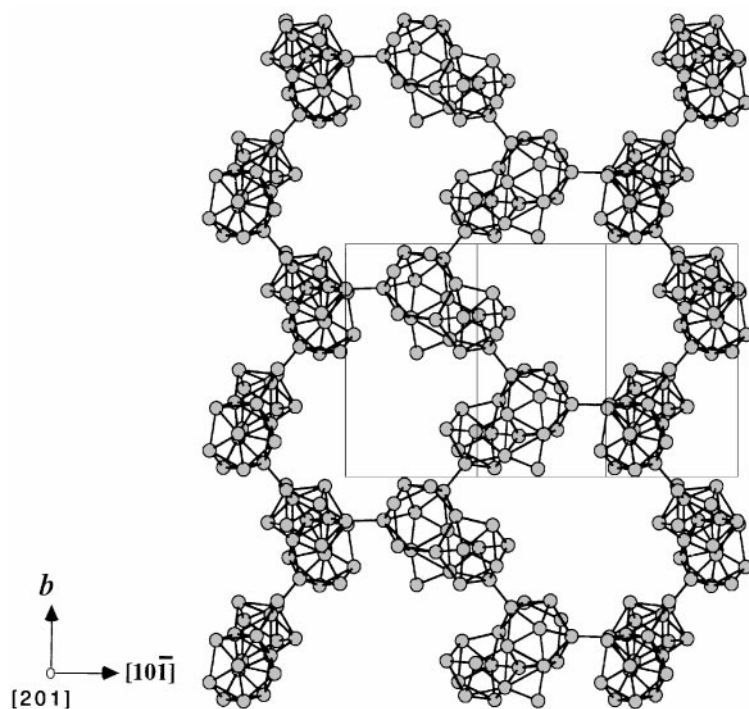


FIG. 9. Arrangement of B_{20} units in the α - AlB_{12} structure as projected along the (201) axis onto the (101) plane. The (101) unit plane indicated around the center of figure is divided into three equal parts, each of which corresponds to the d_{102} spacing.

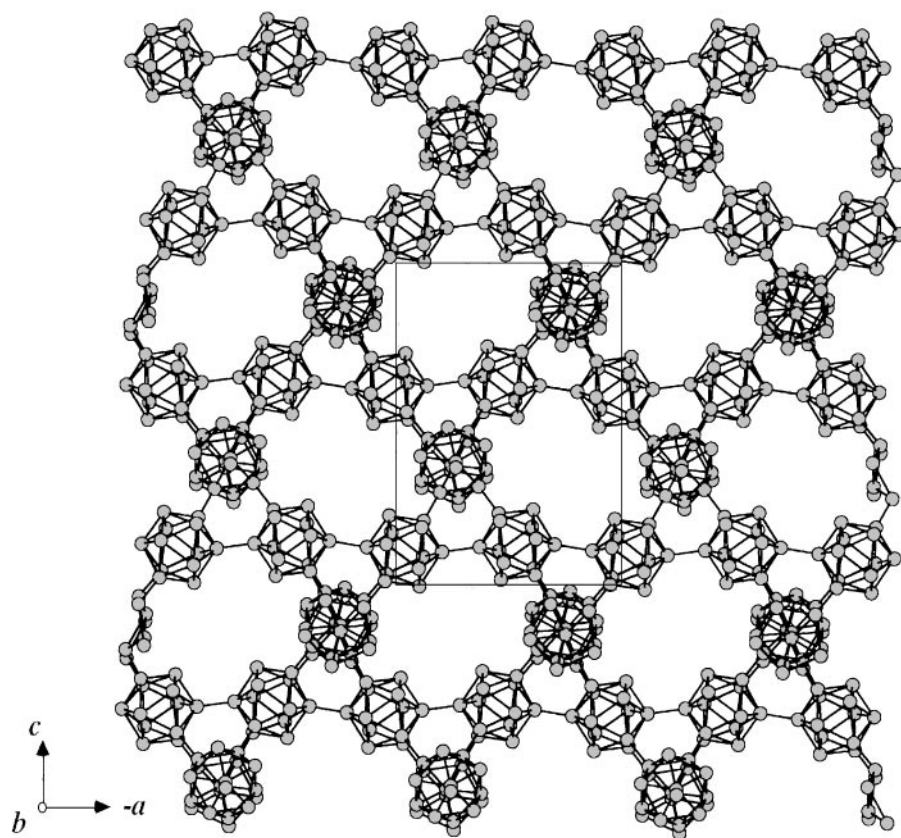


FIG. 10. Arrangement of B_{12} icosahedra in the α - AlB_{12} structure as seen along the b axis.

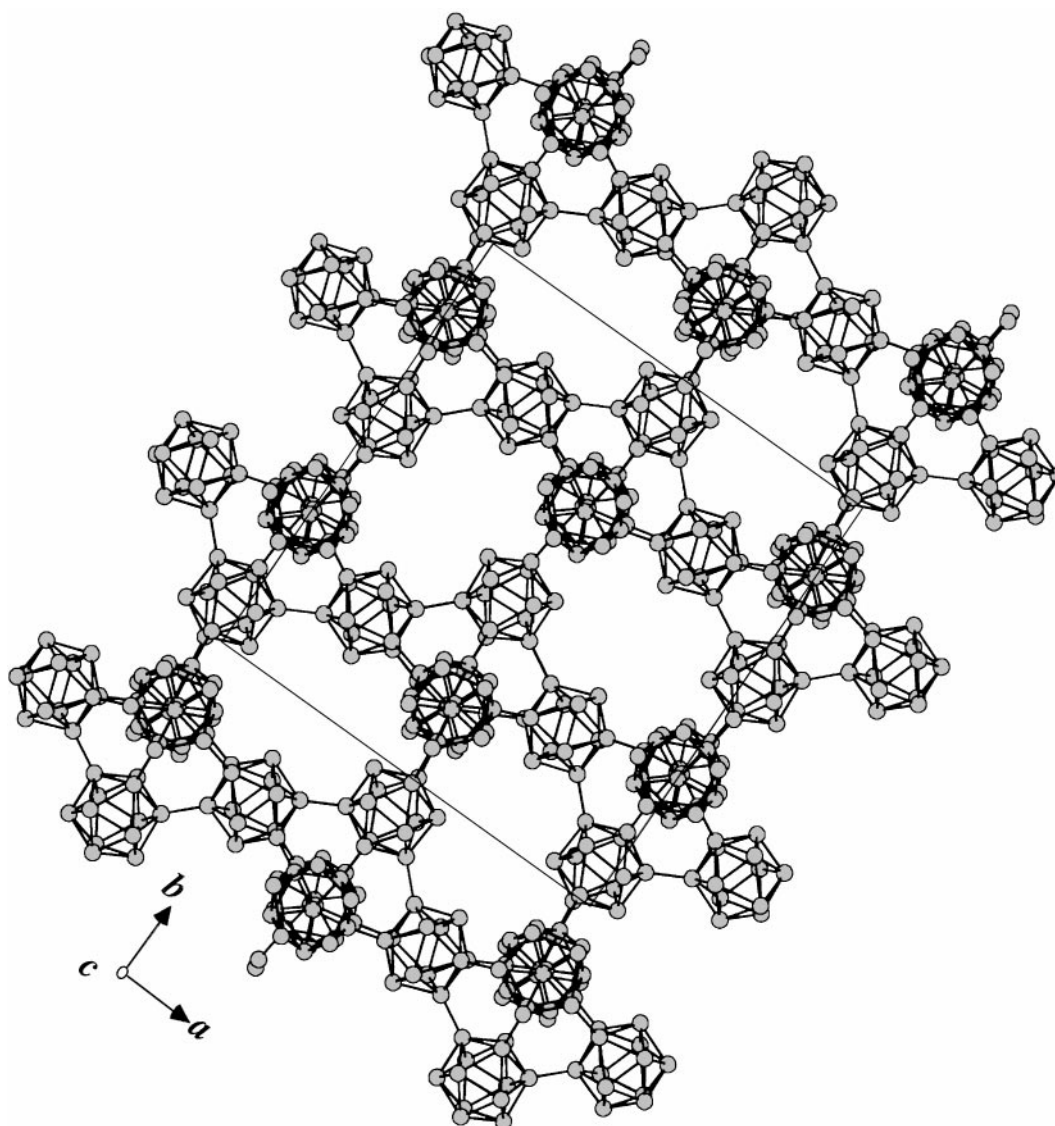


FIG. 11. Arrangement of B₁₂ icosahedra in the γ -AlB₁₂ structure as seen along the c axis.

B₂₀ framework consists of both the B₂₀-(C₂) and B₂₀-(C_s) units. In the intergrown crystal with the relationship shown in Fig. 7, the structure of α -AlB₁₂ is constructed by the translation of the B₁₂ and B₂₀ networks (Figs. 8 and 9) as $(1/2)a_\gamma + (1/3)b_\gamma$, as well as by the unit translation of the networks along the a_x or c_x axis. On the other hand, the structure of γ -AlB₁₂ is generated by operating the 2₁ symmetry operators as indicated in Fig. 7. In the γ -AlB₁₂ structure, consequently, the layers (B₁₂ and B₂₀ networks) with an odd number are those which are generated by the 180° rotation of the basic layer (0-layer or a layer of even number) around the 2₁ axis. In the α -AlB₁₂ structure, on the other hand, all the layers lie at crystallographically equivalent planes. This is the fundamental difference in the struc-

tures between α -AlB₁₂ and γ -AlB₁₂, together with the fact that the B₂₀ framework in the former consists of only B₂₀-(C₂) units, while that in the latter consists of both types of B₂₀ units.

An icosahedral B₁₂ network in α -AlB₁₂ (viewed down the b axis) and in γ -AlB₁₂ (viewed down the c axis) is shown in Figs. 10 and 11, respectively; they are both parallel with the (010)_x or (001)_y plane of the intergrown crystal as shown in Fig. 7. On first viewing, they might seem to have the same icosahedral B₁₂ arrangement. In the α -AlB₁₂ structure, however, there are infinite linear chains of icosahedral B₁₂ units running along the a axis as seen in Fig. 3, although in the γ -AlB₁₂ structure there are no such infinite linear chains running along the corresponding direction.

The difference between the arrangements of B_{12} icosahedral units in these networks reflects the difference of the unit plane areas parallel to these networks (Fig. 7).

As mentioned in the Introduction, the γ - AlB_{12} crystal phase in the pure binary Al-B system displays two unusual natures: First, it is almost always obtained as syntactically intergrown crystals with α - AlB_{12} ; and second, by pulverization (to prepare a sample for powder X-ray diffraction), the γ - AlB_{12} phase in the intergrown crystals is transformed into the α - AlB_{12} phase. The close similarity of the B_{12} and B_{20} networks in α - AlB_{12} (Figs. 8 and 9) with those in γ - AlB_{12} (Figs. 5 and 6), which are all parallel with the intergrowth plane $(101)\alpha$ or $(100)\gamma$, may facilitate the formation of the syntactically intergrown crystals.

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